



5. If the reaction is not complete, continue to add acetic anhydride (649.37 g or 601.3 mL) until TLC shows full consumption of the starting material.
6. Once the reaction is complete, remove the stir bar.
7. Place the reaction mixture into the rotoevaporator flask.
8. Evaporate all solvents. (You will know when this is done...it will not smell like vinegar).
9. Weight out the round bottomed flask and calculate the percent yield.
10. Store the product in a cool, dark space.
11. To verify the product, send off for Mass Spectrometry, or Liquid Chromatography—  
Mass Spectrometry.

### Important equations:

$$\text{Theoretical Yield} = \left( \frac{\text{grams of THCa}}{\text{mass of THCa}} \right) (\text{Mass of product})$$

$$\text{Percent Yield} = \left( \frac{\text{Actual Yield}}{\text{Theoretical Yield}} \right) 100\%$$

### Safety Data Sheets:

Acetic Anhydride

[https://www.sigmaaldrich.com/Graphics/COFAInfo/SigmaSAPQM/SPEC/24/242845/242845-BULK\\_SIAL\\_.pdf](https://www.sigmaaldrich.com/Graphics/COFAInfo/SigmaSAPQM/SPEC/24/242845/242845-BULK_SIAL_.pdf)

Ethyl Acetate

[https://www.sigmaaldrich.com/Graphics/COFAInfo/SigmaSAPQM/SPEC/31/319902/319902-BULK\\_SIGALD\\_.pdf](https://www.sigmaaldrich.com/Graphics/COFAInfo/SigmaSAPQM/SPEC/31/319902/319902-BULK_SIGALD_.pdf)